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Lavoisier was a French chemist,  
philosopher, and father of modern

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chemistry. Lavoisier studied combustion and oxygen. His new system of chemistry led to a complete revision in the classification of elements and in chemical nomenclature. The scientist was arrested during the revolution and executed. Biography

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"It will no doubt be a matter of surprise," Lavoisier writes in the Preface to his Elements of Chemistry, "that in a treatise upon the elements of chemistry, there should be no chapter on the constituent and elementary parts of matter; but I shall take occasion, in this place, to remark that the fondness for reducing all the bodies in nature to three or four elements, proceeds from a prejudice

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Lavoisier on Elements - Yale University



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It is generally accepted that Lavoisier's great accomplishments in chemistry stem largely from his changing the science from a qualitative to a quantitative one. Lavoisier is most noted for his discovery of the role oxygen plays in combustion. He recognized and named oxygen (1778) and hydrogen (1783), and opposed the phlogiston theory.

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Many scientists of the day refused to accept Lavoisier's theories but his book, The Elements of Chemistry, silenced the doubters. It was published in 1789, a significant year. Having shown that water and air were not chemical elements Lavoisier included in his book a list of those substances that he believed to be elementary.

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